

SOKOL, V.I.; TOKAREVA, S.A.; SOKOVNIN, Ye.I.

Determination of the density and refractive indices of
sodium and potassium ozonides. Izv. AN SSSR. Ser. khim.
no.12:2220-2221 D '63. (MIRA 17:1)

1. Institut obshchey i neorganicheskoy khimii im. N.S.
Kurnakova AN SSSR.

TOKAREVA, S. A.; PILIPENKO, G. P.

Thermal decomposition of sodium ozonide. Izv AN SSSR Ser Khim
no. 4:740-743 Ap '64. (MIRA 17:5)

1. Institut obshchey i neorganicheskoy khimii im. N. S.
Kurnakova AN SSSR.

L 21498-66 EWT(m)/EWP(t) LJP(c) JD/JG

ACC NR: AP6009438

SOURCE CODE: UR/0075/66/021/003/0380/0381

AUTHOR: Tokareva, S. A.; Pilipenko, G. P.

ORG: Institute of General and Inorganic Chemistry im. N. S. Kurnakov, AN SSSR,
Moscow (Institut obshchey i neorganicheskoy khimii AN SSSR)

TITLE: Chemical analysis of sodium ozonide

SOURCE: Zhurnal analiticheskoy khimii, v. 21, no. 3, 1966, 380-381

TOPIC TAGS: analytical chemistry, quantitative analysis, gravimetric analysis, gas
volumetric analysis, ozonide, sodium compound, sodium ozonide

ABSTRACT: A combined gas-volumetric and gravimetric method has been developed for
the quantitative analysis of sodium ozonide with indirect sampling at -50 to -60C
because of the instability of sodium ozonide at room temperature. Two samples were
required for a complete analysis. One sample was used for determining superoxidic
(active) oxygen by the combined thermal and aqueous decomposition of ozonide in an
apparatus which was described. The second sample was used for determining total
Na₂O content as sodium sulfate, after decomposing ozonide as described. Accuracy
of the method was ±1%, as determined by analyzing potassium ozonide by this and an-
other [unspecified] method. The suggested method can be used for the analysis of
other ozonides unstable at room temperature. Orig. art. has: 2 figures and 1 table.

[JK]

SUB CODE: 07/ SUBM DATE: 26Feb65/ ATD PRESS: 4022
Card 1/1 UDC: 543.70

5.2100

29530
S/078/61/006/011/005/013
B101/B147

AUTHORS: Tsentsiper, A. B., Tokareva, S. A.

TITLE: Interaction of carbon monoxide with sodium and potassium peroxide

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 6, no. 11, 1961, 2474-2480

TEXT: Since no data are available on the reactivity of peroxides of alkali metals, the reaction of NaO_2 and KO_2 with CO and $\text{CO} + \text{H}_2\text{O}$ was studied. [Abstracter's note: Compounds NaO_2 and KO_2 should be better termed dioxides.] The reaction flask containing 0.5 - 0.8 g of alkali dioxide was evacuated to 10^{-4} mm Hg and heated either in a TC-15 (TS-15) thermostat at 95°C , or in a glycerol bath at 140, 160, or 180°C . Then, CO or $\text{CO} + \text{H}_2\text{O}$ were supplied. The changes in pressure were measured by means of a butyl-phthalate differential manometer (15 mm dibutyl phthalate = 1 mm Hg). NaO_2 (86.2%) and KO_2 (92.6%) were used as initial products. Impurities consisted of peroxide (Me_2O_2), carbonate, and hydroxide. CO

Card 1/8 4

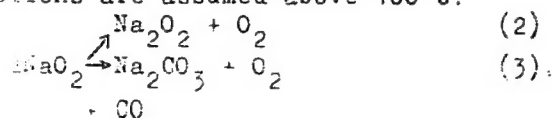
29530

S/078/61/006/011/005/013

B101/B147

Interaction of carbon monoxide...

was synthesized from 85% HCOOH and H₂SO₄ at 80°C. The total oxygen of the solid phase was determined by decomposing with 0.5% CuSO₄ solution; active O₂ (peroxide O₂) was determined by titration with 0.1 N KMnO₄; dioxide O₂ was calculated from the difference. The total alkalinity and CO₂ bound as carbonate were titrimetrically determined (difference of equivalence points for phenolphthalein and methyl orange serving as indicators). The analytical error was <1%. Investigation of thermal stability of NaO₂ at the temperatures mentioned showed that NaO₂ was stable up to 95°C. Only at higher temperatures, the reaction sets in: 2NaO₂ → Na₂O₂ + O₂ (1). Data for the reaction with dry CO are given in Table 2. For NaO₂, the following reactions are assumed above 100°C:



Card 2/4

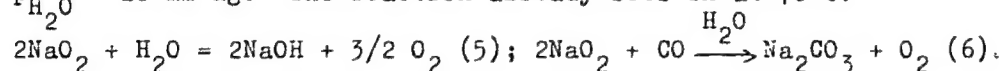
Interaction of carbon monoxide...

29530
S/078/61/006/011/005/013
B101/B147

KO₂ reacts with CO already at 95°C: $2\text{KO}_2 + \text{CO} = \text{K}_2\text{CO}_3 + \text{O}_2$ (4). Since KO₂ only decomposes at 400°C, this reaction may be due to the higher reactivity of KO₂. In the beginning, the intermediate complex KO₂·CO forms. This absorption of CO causes an initial fall in pressure in the apparatus. The reaction with CO and H₂O vapor was studied at

$p_{\text{CO}} = 6 - 79$ mm Hg and $p_{\text{H}_2\text{O}} = 11$ mm Hg, and at $p_{\text{CO}} = 11 - 40$ mm Hg and

$p_{\text{H}_2\text{O}} = 20$ mm Hg. The reaction already sets in at 70°C:



Formation of carbonate proceeds over a stage catalyzed by NaOH: $\text{CO} + \text{O} = \text{CO}_2$ (7) and $\text{CO}_2 + 2\text{NaOH} \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O}$ (8). In the absence of NaOH, reaction Eq. (7) does not take place. Experiments with high p_{CO} showed a lower degree of conversion of NaO₂, since in this case H₂O diffusion to the NaO₂ surface was inhibited and reaction Eq. (6) was

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Interaction of carbon monoxide...

29530
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suppressed. A paper by T. V. Rode, G. A. Gol'der (Izv. AN SSSR, Otd. khim. n., 299 (1956)) is mentioned. There are 3 figures, 3 tables, and 11 references: 5 Soviet and 6 non-Soviet. The two most recent references to English-language publications read as follows: P. Gills, J. Margrave, J. Phys. Chem., 60, 1334 (1956); E. Neuman, J. Chem. Phys., 2, 31 (1954). X

SUBMITTED: September 15, 1960

Table 2. Interaction of NaO_2 and KO_2 with carbon monoxide

Legend: (a) temperature, $^{\circ}\text{C}$; (b) initial pressure, mm Hg; (c) duration of experiment, min; (d) composition of the end product, % by weight; (e) total degree of conversion, %; (f) degree of conversion up to Na_2O_2 , %; (g) degree of conversion up to Na_2CO_3 , %.

Card 4/64

ACCESSION NR: AP4019271

S/0192/64/005/001/0142/0144

AUTHORS: Kuznetsov, V.G.; Bakulina, V.M.; Tokareva, S.A.;
Zimina, A.N.

TITLE: X ray study of sodium ozonide, NaO sub 3

SOURCE: Zhurnal strukturnoy khimii, v. 5, no. 1, 1964, 142-144

TOPIC TAGS: x ray study, sodium ozonide, symmetry, cell dimension,
interplaner distance, volume centered tetragonal lattice, sodium,
sodium compound

ABSTRACT: Sodium ozonide was obtained by reaction of ozone with
dehydrated sodium hydroxide at -80C for 3 hrs. with subsequent ex-
traction from liquid ammonia. The solvent was removed in a vacuum
at -50C. The crystalline product contained 85% sodium ozonide.
Specimens of sodium ozonide synthesized at a temperature interval
of 0 to 5C and separated by subsequent extraction with liquid
ammonia were studied simultaneously. From X-ray photographs it was

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ACCESSION NR: AP4019271

possible to measure more lines and obtain more accurate values, and also to determine the symmetry and cell dimensions. Indexing of x-ray photographs by means of Helly's curves provided better agreement of measured and calculated interplaner distances for a volume centered tetragonal lattice with the ratio $c/a = 0.66$ and with periods $a = 11.65$ and $c = 7.66$ Å. Deviation is observed for the first diffuse line with $d = 3.927$ Å, which is explained by a large error of measurement for this line. The density of sodium ozonide found by the hydrostatic suspension method, is 1.6 g./cm^3 . The number of molecules in the unit cell is 14. As a result of analysis of extinction and of value $N=14$, spatial group I of 4ttt was tentatively selected. Orig. art. has: 1 table, 1 figure.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii im. N.S. Kurnakova AN SSSR (Institute of General and Inorganic Chemistry AN SSSR)

SUBMITTED: 19Jun63

DATE ACQ: 27Mar64

ENCL: 00

SUB CODE: CH

NO REF SOV: 005

OTHER: 003

Card 2/2

ACCESSION NR: AP4033392

S/0062/64/000/004/0740/0743

AUTHOR: Tokareva, S. A.; Pilipenko, G. P..

TITLE: Thermal decomposition of sodium ozonide

SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 4, 1964, 740-743

TOPIC TAGS: sodium ozonide, thermal decomposition, thermal stability, rate of decomposition, synthesis

ABSTRACT: The thermal stability of sodium ozonide at different temperatures in the absence of moisture was studied. The sodium ozonide used was synthesized by ozonizing NaOH at -80 to -100C, extracting with liquid ammonia and removing the latter under vacuum at -50C. Curves for the conversion of NaO_3 (see fig. 1 of the enclosure) and its rate of decomposition at -20, -10, 0 and 10C were constructed from weighings on a McBain balance. At -30C there was no weight change after 4 hours at 10^{-1} mm Hg. At -20C the decomposition is slow, and at -20 and -10C the NaO_3 decomposed to NaO_2 . At temperatures above 0C the thermal decomposition of NaO_3 is accompanied by the reaction of the formed NaO_2 with H_2O to give NaOH. Orig. art. has: 3 figures and 1 table.

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ACCESSION NR: AP4033392

ASSOCIATION: Institut obshchey i neorganicheskoy khimii im. N. S. Kurnakova,
Akademii nauk SSSR (Institute of General and Inorganic Chemistry, Academy of
Sciences, SSSR)

SUBMITTED: 11Mar63

DATE ACQ: 15May64

ENCL: 01

SUB CODE: IC

NO REF SOV: 002

OTHER: 001

Card - 2/3

ACCESSION NR: AP4033392

ENCLOSURE: 01

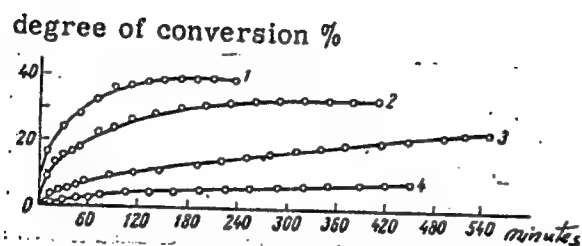


Fig. 1. Relationship between the degree of sodium ozonide conversion and time at different temperatures. 1-- 10C, 2--0C, 3-- -10C, 4-- -20C.

Card 3/3

GEL'PERIN, N.I.; NATRADZE, A.G.; TOKAREVA, S.A.

Continuous-process production of barium sulfate. Khim. i med.
no. 12:18-26 '59. (MIRA 13:10)

(BARIUM SULFATE)

GEL'PERIN, N.I.; NATRADZE, A.G.; TOKAREVA, S.A.

Improvement in the method of obtaining santonin. Khim. i med.
no. 12:33-40 '59. (MIRA 13:10)

(SANTONIN)

10.11.11, 5.11
MIRONOV, K.Ye.; PRONINA, M.Z.; TOKAREVA, S.A.

Study of crystallization area of H_2O_2 - $NaClO_4$ - H_2O and H_2O_2 - $LiClO_4$ - H_2O systems. Zhur. neorg. khim. 3 no.2:508-516 1958.

(MIRA 11:4)

1. Institut obshchey i neorganicheskoy khimii im. N.S. Kurnakova
AN SSSR laboratoriya perekisnykh soedineniy.

(Sodium salts) (Lithium salts)

(Phase rule and equilibrium)

Tokareva, S.A.

AUTHORS: Mironov, K. Ye., Pronina, M. Z., Tokareva, S. A. 78-2-37/43

TITLE: An Investigation of Crystallization in the Systems
 $H_2O_2-NaClO_4-H_2O$ and $H_2O_2-LiClO_4-H_2O$
 (Izucheniye poverkhnosti kristallizatsii sistem $H_2O_2-NaClO_4-H_2O$ i $H_2O_2-LiClO_4-H_2O$)

PERIODICAL: Zhurnal Neorganicheskoy Khimii, 1958, Vol. 3, Nr 2,
 pp. 508-516 (USSR)

ABSTRACT: A complete investigation of the diagrams of $H_2O_2-NaClO_4-H_2O$ and $H_2O_2-LiClO_4-H_2O$ was performed. The concentration of H_2O_2 was obtained by repeated distillation in a vacuum with a purity of 99,8%. The formation of crystals occurs at deep undercooling (60-70°C lower than the equilibrium of the crystallization). In the system $H_2O_2-NaClO_4-H_2O$ the following phases occur: $H_2O_2 \cdot 2H_2O$, ice, $NaClO_4 \cdot H_2O$ and $NaClO_4$. In the system $LiClO_4-H_2O$ the following phases are obtained at 0°C: ice, $LiClO_4 \cdot 3H_2O$, $LiClO_4 \cdot H_2O$ and $LiClO_4$.

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An Investigation of Crystallization in the Systems

78-2-37/43

H_2O_2 - $NaClO_4$ - H_2O and H_2O_2 - $LiClO_4$ - H_2O

In the liquidus of the binary system H_2O_2 - $LiClO_4$, $LiClO_4$ and H_2O_2 develop. In the ternary system H_2O_2 - $LiClO_4$ - H_2O the following phases are produced: ice, H_2O_2 , $H_2O_2 \cdot 2H_2O$, $LiClO_4$, $LiClO_4 \cdot H_2O$ and $LiClO_4 \cdot 3H_2O$.

From these results follows that no peroxyhydrates of sodium- and lithium perchlorate are produced in the binary systems H_2O_2 - $NaClO_4$ and H_2O_2 - $LiClO_4$.

Under the influence of aqueous solutions of H_2O_2 upon the perchlorates of sodium and lithium the authors obtained hydrate forms of perchlorates. $LiClO_4$ hydrated especially intensively. There are 8 figures, 5 tables, and 9 references, 7 of which are Slavic.

ASSOCIATION: Institute for General and Anorganic Chemistry imeni N. S. Kurnakov AS USSR (Institut obshchey i neorganicheskoy khimii imeni N. S. Kurnakova Akademii nauk SSSR)
Laboratory for Peroxy-Compounds (Laboratoriya perekisnykh soedineniy)

SUBMITTED: February 19, 1957
Card 2/3

An Investigation of Crystallization in the Systems
· H_2O_2 - $NaClO_4$ - H_2O and H_2O_2 - $LiClO_4$ - H_2O)

78-2-37/43

AVAILABLE: Library of Congress

Card 3/3

S/078/62/007/005/003/014
B101/B110

AUTHORS: Kuznetsov, V. G., Tokareva, S. A., Dobrolyubova, M. S.

TITLE: X-ray diffraction analysis of sodium ozonide NaO_3

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 5, 1962, 967 - 970

TEXT: X-ray powder patterns were taken for determining the crystallization form of NaO_3 . NaO_3 was synthesized by reaction of O_3 with anhydrous NaOH at -80°C and subsequent extraction with liquid NH_3 which was removed in vacuo at -50°C . The resulting product (red crystals) contained 90-92% NaO_3 . Because of the instability of NaO_3 , the x-ray patterns were taken at nitrogen temperature by an YPC-55 (URS-55) camera. The x-ray patterns of NaNO_3 , $(\text{NH}_4)\text{NO}_3$, NaOH , $\text{NaOH}\cdot\text{H}_2\text{O}$, and NaO_2 were taken for comparison. NaO_3 was found to contain small amounts of NaOH and NaO_2 . The indication of the x-ray patterns showed good agreement with the interplanar spacings

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X-ray diffraction analysis of sodium ...

S/078/62/007/005/003/014
B101/B110

calculated for a tetragonal cell, and the lattice data were $a = 10.43$, $c = 6.88\text{kX}$; $c/a = 0.660$. Nearly all lines of the x-ray patterns can be explained by superposition of reflections with different hkl indices. In default of systematic extinctions it was not possible to determine the space group. A comparison of NaO_3 with KO_3 data (G. S. Zhdanov, Z. V. Zvonkova, Zh. fiz. khimii, 25, 100 (1951)) showed no isomorphism between NaO_3 and KO_3 . The x-ray pattern of NaO_3 is also different from the patterns of NaN_3 and NaNO_2 . There are 1 figure and 1 table. The most important English-language references are: A. D. McLachlam, M. C. R. Symons, M. G. Townsend, J. Chem. Soc., 952 (1959); I. J. Solomon, A. I. Kacmaber, J. Phys. Chem., 64, 168 (1960).

SUBMITTED: May 8, 1961

Card 2/2

CHERNOMORDIKOV, V. V.; Prinimali uchastiye: GORYACHEVA, M., student-diplomnik; TOKAREVA, T., student-diplomnik; CHERNYSHEVA, Ye., student-diplomnik; SHUTOVA, M., student-diplomnik; MAMATKINA, E., studentka

Thermophily and hygrophily of Norway and black rats. Nauch. dkl. vys. shkoly; biol. nauki no.3:69-72 '62. (MIRA 15:7)

1. Kafedra zoologii pozvonochnykh Moskovskogo gosudarstvennogo universiteta im. M. V. Lomonosova (for Goryacheva, Tokareva, Chernysheva, Shutova). 2. Moskovskiy zaochnyy sel'skokhozyaystvennyy institut (for Mamatkina).

(RATS) (ZOOLOGY-ECOLOGY)

TOKAREVA, T.G.

Morphological and histochemical changes in the tissues of lymphoid organs in intrainestinal immunization. Vest. AMN SSSR 19 no.12:61-65 '64. (MIRA 18:4)

1. Moskovskiy nauchno-issledovatel'skiy institut vaktsin i syvorotok imeni Mechnikova Ministerstva zdravookhraneniya SSSR.

MESHALOVA, A.N.; DROZDOVA, V.N.; TOKAREVA, T.G.

Comparative study of the immunological reactivity of the
organism in enteral and subcutaneous immunization against
typhoid fever. Zhur. mikrobiol., epid. i immun. 42 no.7:52-57
Jl '65. (MIRA 18:11)

1. Moskovskiy institut vaktsin i sывorotek imeni Mechnikova.

ATYAKIN, A.K.; VOLOKITENKOV, A.A.; LITVINOV, N.N.; TOKAREVA,
T.N., ved. red.; YASHCHURZHINSKAYA, A.B., tekhn. red.

[Testing and drilling exploratory boreholes under
complicated conditions] Oprobovanie i burenie razvedoch-
nykh skvazhin v oslozhnennykh usloviakh. Leningrad,
Gostoptekhzdat, 1963. 189 p. (MIRA 17:2)

YAKUTSENI, Vera Prokof'yevna; TOKAREVA, T.N., vedushchiy red.; DEM'YANENKO, V.I., tekhn.ted.

[Characteristics of the formation of helium-bearing gas fields; prospecting methods.] Zakonomernosti formirovaniia zalezhei geli-
enosnykh gazov; k metodike poiskov. Leningrad, Gostoptekhizdat,
1963. 130 p. (Leningrad. Vsesoiuznyi neftianoi nauchno-issledo-
vatel'skii geologorazvedochnyi institut. Trudy, no.222).

(MIRA 17:2)

ANDREYEV, Boris Aleksandrovich; KLUSHIN, Igor' Gennad'yevich;
SEMENOV, A.S., retsenzent; MIRONOV, V.S., retsenzent;
DEMENITSKAYA, R.M., doktor geol.-miner. nauk, retsenzent;
MIKHAYLOV, N.N., nauchnyy red.; TOKAREVA, T.N., ved. red.;
SAFRONOVA, I.M., tekhn. red.

[Geological interpretation of gravity anomalies]Geologicheskoe istolkovanie gravitatsionnykh anomalii. Leningrad, Gostoptekhizdat, 1962. 495 p. (MIRA 16:3)
(Gravity anomalies)

TEST, B.I.; OSIPOVA, Z.V.; SYCHEV, V.Ya.; SOROKOV, D.S., nauchnyy red.;
TOKAREVA, T.N., vedushchiy red.; SAFRONOVA, I.M., tekhn.red.

[Mesozoic sediments of the Zhigansk region] *Mezozoiskie*
otlozheniia Zhiganskogo raiona. Leningrad, Gos. nauchn.-tekhn.
izd-vo nef. i gorno-topl. lit-ry, Leningr. otd-nie, 1962.
117 p. (Leningrad. Nauchno-issledovatel'skii institut geologii
arktiki. Trudy, vol. 131). (MIRA 15:11)
(Verkhoyansk Range—Geology)

LAVROVA, M.A., red.; FADDEYEVA, A.P., red.; ZHINGAREVA--
DOBROSEL'SKIY, A.T., red.; TOKAREVA, T.N., ved. red.

[Problems of the stratigraphy of Quaternary sediments
in the northwestern area of the European part of the
U.S.S.R.] Voprosy stratigrafii chetvertichnykh otlozhenii
Severo-Zapada Evropeiskoi chasti SSSR; sbornik statei.
Leningrad, Gostoptekhizdat, 1962. 198 p. (MIRA 18:5)

1. Nauchno-tekhnicheskoye gornoye obshchestvo, Moscow.
Leningradskoye oblastnoye upravleniye.

L 3381-66 EWT(m)/EWP(j)/T RM

ACCESSION NR: AP5022093

UR/0138/65/000/008/0042/0044

678.06:685.314.33.002.2

AUTHOR: Tokareva, T. Ye.; Snitsarenko, L. G.; Volkova, N. A.; Baksht, O. V.;
Zel'dich, E. I.; Kheyfets, F. M.

TITLE: Compounding and technology for manufacturing winter-proof boots

SOURCE: Kauchuk i rezina, no. 8, 1965, 42-44

TOPIC TAGS: rubber chemical, antifreeze, synthetic material, butadiene styrene rubber, filler, plasticizer, thermoelasticity, special purpose clothing, rubber/SKMS-10 rubber

ABSTRACT: Formulations and technology for making frost-resistant boots which retained their elasticity at -50C were worked out and introduced commercially. Formulations for all parts except the tricot-backed boot tops were based on frost resistant rubber SKMS-10, and natural rubber was used in formulation for fabric application. The antifreeze effectiveness of dibutylphthalate, dibutylsebacinate, MVP oil, "plasticizer" oil and transformer oil was evaluated. The first two compounds gave the best frost-resistance at -50 C, and formulations containing dibutylphthalate had the greatest resistance to aging and became brittle below

Card 1/2

L 3381-66

ACCESSION NR: AP5022093

-65C . Different types of carbon black had little effect on frost-resistance. Manufacturing technology for making frost-resistant regular and fisherman's boots is analogous to that for making ordinary molded boots. Orig. art. has: 2 tables

ASSOCIATION: Nauchno-issledovatel'skiy institut rezinovykh i lateksnykh izdeliy (Scientific Research Institute for Rubber and Latex Products); Zavod "Krasnyy bogatyr" (Krasnyy Bogatyr Plant)

SUBMITTED: 00

ENCL: 00

SUB CODE: MT, IE

NR REF SOV: 005

OTHER: 000

Card 2/2

POZIN, A.A.; TOKAREVA, T.Ye.; KOCHKIN, K.I.; PYATETSKAYA-SHAPIRO, I.P.

Mechanized method for the manufacture of warm rubber boots. Kauch.
i rez. 24 no.4:32-35 Ap '65. (MIRA 18:5)

1. Nauchno-issledovatel'skiy institut rezinovykh i lateksnykh
izdeliy.

TOKAREVA, T.Ye.; SNITSARENKO, L.G.; VOLKOVA, N.A.; BAKSHT, O.V.;
ZEL'DICH, E.I.; KHEYFETS, F.M.

Formulas and technology for the manufacture of frost-resistant
boots. Kauch. i rez. 24 no.8:42-44 '65.

(MIRA 18:10)

1. Nauchno-issledovatel'skiy institut rezinovykh i lateksnykh
izdeliy i zavod "Krasnyy bogatyr'.

TOKAREVA, V.A., kand.med.nauk; GEYMOS, Ye.K., vrach

Pemphigus and its treatment. Oft. zhur. 16 no.1:3-7 '61.
(MIRA 14:3)

1. Iz kafedry glaznykh bolezney (zav. - prof. N.A.Pletneva) II
Moskovskogo meditsinskogo instituta i 1-y detskoy klinicheskoy
bol'nitsy.

(PEMPHIGUS)

ALOV. A.A.; TOKAREVA, V.A., redaktor; BUKHVALOVA, K.I., redaktor.

[Electrodes for arc welding and weld deposition] Elektrody dlia dugo-
voi svarki i naplavki. Sverdlovsk, Gos. nauchno-tekhn. izd-vo mashino-
stroit. lit-ry [Sverdlovskoe otd-nie] 1947. 86 p. (MLRA 7:7)
(Welding)

TOKAREVA, V.D.

Cooling the journal bearings of cement mills. TSement 27 no.3:
26 My-Je '61. (MIRA 14:7)

1. Bryanskiy tsementynyy zavod.

(~~Byanskiy~~ Cement plants--Equipment and supplies)
(Bearings (Machinery))

5(3)
 SOV/79-29-9-65/76
 AUTHORS: Skvortsova, N. I., Tokareva, V. Ya., Belov, V. N.
 TITLE: Synthesis of Nerolidol, Methyl Nerolidol, Farnesal and Methyl Farnesal
 PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 9, pp 3113-3117(USSR)
 ABSTRACT: Among the investigations published in recent years the synthesis of the compounds with terpenoid structure, which are important for perfumery, was carried out by using geranyl chloride (I) (8-chloro-2,6-dimethyl octadiene-2,6) and methyl geranyl chloride (II) (8-chloro-2,3,6-trimethyl octadiene-2,6) as intermediate products (Refs 1-5). Since there are good methods of synthesizing these chlorides and since they will be industrially produced in the near future the authors used chloride (I) and (II) for the synthesis of the following compounds: nerolidol (V) (2,6,10-trimethyl dodecatriene-2,6,11-ol-10), methyl-nerolidol (VI) (2,3,6,10-tetramethyl dodecatriene-2,6,11-ol-10), farnesal (VII) (2,6,10-trimethyl dodecatriene-2,6,10-al-11), methyl farnesal (VIII) (2,3,6,10-tetramethyl dodecatriene-2,6,10-al-11) (Reaction Scheme). The formation of geranyl acetone (III) (2,6-dimethyl undecadiene-2,6-on-10) by reacting geranyl chloride with acetic acid ester is described in publications (Refs 5, 6). In the present paper the synthesis of geranyl

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SOV/79-29-9-65/76

Synthesis of Nerolidol, Methyl Nerolidol, Farnesal and Methyl Farnesal

acetone and methyl geranyl acetone (IV) (2,3,6-trimethyl undecadiene-2,6-on-10) was carried out without separation of the substituted acetoacetic ester from the reaction mixture. The transformation of the ketones (III) and (IV) into the tertiary alcohols (V) and (VI) was made by reacting these ketones with vinyl magnesium bromide according to H. Normant (Ref 7). The transition from the tertiary alcohols (V) and (VI) into the aldehydes (VII) and (VIII) took place via the alkyl regrouping and the oxidation according to the method generally used for such transformations (Ref 6). The constants of the synthesized nerolidol and farnesal samples agree with those given in publications. Methyl nerolidol and methyl farnesal have hitherto been unknown. There are 12 references, 6 of which are Soviet.

ASSOCIATION: Nauchno-issledovatel'skiy institut sinteticheskikh i natural'-nykh dushistykh veshchestv (Scientific Research Institute of Synthetic and Natural Aromatic Substances)

SUBMITTED: August 25, 1958

Card 2/2

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Instituta im Pastera (direktor F.I.Krasnik) i iz kafedry patologiche-
skoy anatomii Leningradskogo veterinarnogo instituta (zav. - prof.
V.Z.Chernyak)

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Country : USSR
Category: Virology. Viruses of Man and Animals.
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Abs Jour: Ref Zhur-Biol., No 23, 1958, No 103569

Author : Tokarevich, K. M.

Inst :

Title : Recurrent Typhus

Orig Pub: Sb. Rikettsiozy, Leningrad, 1958, 5-14.

Abstract: No abstract.

Card : 1/1

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Country : USSR
Category: Virology. Viruses of Man and Animals.
Rickettsias.

Abs Jour: Ref Zhur-Biol., No 23, 1958, No 103570

Author : Tokarevich, K.N.; Epstein, Ye. F.; Klushina, T.A.
Inst : -
Title : Some Results of Detection of Atypical Forms of Typhus

Orig pub: Sb. Rikettsiozy, Leningrad, 1958, 42-50.

Abstract: No abstract.

Card : 1/1

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Country : USSR
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Author : Tokarevich, K.N.
Inst : _____
Title : On So-Called Intrahospital Typhus Infections

Orig Pub: Sb. Rikettsiozy. Leningrad, 1958, 92-98

Abstract: No abstract.

Card : 1/1

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Rickettsias.

Abs Jour: Ref Zhur-Biol., No 23, 1958, No 103573

Author : Tokarevich, K.N.
Inst : -
Title : Recurrent Typhus and So-Called Brill's Disease

Orig Pub: Sb. Rikettsiozy. Leningrad, 1958, 99-120

Abstract: No abstract.

Card : 1/1

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Country : USSR
Category: Virology. Viruses of Man and Animals.
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Abstr Jour: Ref Zhur-Biol., No 23, 1958, No 103574

Author : Tokarevich, K.N.

Inst : -

Title : Those Repeatedly Sick with Typhus as a Source of
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Orig Pub: Sb. Rikettsiozy. Leningrad, 1958, 121-126.

Abstract: No abstract.

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Abs Jour: Ref Zhur-Biol., No 23, 1958, No 103582

Author : Tokarevich, K. N.; Vasil'yeva, L.D.

Inst :

Title : First Cases of "Q" Fever in Leningrad

Orig Pub: Sb. Rikketsiozy. Leningrad, 1958, 182-191

Abstract: No abstract.

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(MIRA 16:7)

(RUSSIA, NORTHWESTERN—LEPTOSPIROSIS)

TOKAREVICH, K. M., VASIL'YENK, L. D., AMOSENKOVA, N. I., DAYTER, A. B.,
POPOVA, E. M.

"Materials for the further study of the local Q-fever focus in
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Preservation of tularemia bacteria during streptomycin therapy.
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KUDRYAVTSEV, P.I., kand.filosof.nauk, red.; TOKAREVICH, K.N., prof..
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1. Leningradskiy institut epidemiologii, mikrobiologii i gigieny
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(LENINGRAD PROVINCE—Q FEVER)

TOKAREVICH, K.N. (Leningrad)

Occupational nature of certain infections with natural endemic areas. Gig. truda i prof. zav. 4 no.12:3-7 D '60. (MIRA 15:3)

1. Institute epidemiologii, mikrobiologii i gigiyeny imeni Pastera i Institut usovershenstvovaniya vrachey imeni S.M. Kirova.

(OCCUPATIONAL DISEASES)

(COMMUNICABLE DISEASES)

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TOKAREVICH, K.N.; VASIL'YEVA, L.D.; AMOSENKOVA, N.I.; DAYTER, A.B.;
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Epidemiology of typhus in the light of the study of the immunological structure of the population with respect to this infection.
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1. Iz Leningradskogo instituta epidemiologii i mikrobiologii imeni Pastera i kafedry epidemiologii Leningradskogo ordena Lenina instituta usovershenstvovaniya vrachey imeni Kirova.
(TYPHUS FEVER) (IMMUNITY)

TOKAREVICH, K.N., prof.

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Instituta epidemiologii i mikrobiologii imeni Pastera i
Voyenno-meditsinskoy ordena Lenina akademii imeni Kirova.

FILATOV, I.F.; TOKAREVICH, K.N.; VISHNYAKOVA, L.A.; FRIDMAN, E.A.

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1. Iz otdela osobo opasnykh infektsiy i laboratorii grippa Leningradskogo instituta epidemiologii i mikrobiologii imeni Pastera.

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